

ORIGINAL RESEARCH PAPER

Application of ZnO/TiO₂ Nanocomposite for the Improvement of Heat Transfer Coefficient in Tube Heat Exchangers

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ABSTRACT

The potential of nanofluids for the improvement of heat transfer coefficient in various heat exchange equipment has been considered and studied as a major application during recent decades. In this research, heat transfer coefficient of ZnO, TiO₂ and ZnO/TiO₂ nanofluids in a shell and tube heat exchanger has been studied experimentally. ZnO nanoparticle was synthesized through precipitation technique. Sol-gel and mechanical techniques were employed to synthesize the ZnO/TiO₂ nano-composite. Particle size analysis (PSA), XRD, FTIR and FE-SEM techniques were used to characterize ZnO/TiO₂ nano-composite. Based on the results, the nanofluid heat transfer coefficient was enhanced by increasing the nanofluid concentration and temperature. The heat transfer coefficient of TiO₂ was higher than that of ZnO nanofluid and the heat transfer coefficients of nano-composites were higher than that of ZnO and TiO₂ nanofluids and the base fluid. Also, the heat transfer coefficient and the overall heat transfer coefficient were increased 3.79 to 9.09 times and 4.27 to 9.14, respectively by increasing the nano-composite content.

Keywords: Heat Transfer Coefficient, Nanocomposite, Nanofluid, Shell and Tube Heat Exchanger

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INTRODUCTION

Heat exchangers are among the most important unit operations in processing industries playing an important role in the management of energy requirements. In order to increase the efficiency and improve the rate of heat transfer in heat exchangers, various techniques such as increasing the heat transfer area and improvement of heat transfer coefficient are proposed. Higher heat transfer coefficient fluids can replace conventional fluids (e.g. water and ethylene glycol), which have led to many research studies during the last decade. Nanofluids are among the very recent alternatives for this purpose. Addition of nano-scale particles into conventional base fluids can cause a significant enhancement in thermal properties [1, 2]. Several Nanoparticles are synthesized and added to various base fluids in order to investigate their effect on

thermal properties such as TiO₂, ZnO, Al₂O₃, and CuO among many others.

Many research activities have been performed on this subject during the last decade. Madhesh et al. investigated the convective heat transfer, pressure drop, friction factor and the rheological properties of Cu-TiO₂ nanofluid in a heat exchanger [3]. The results showed an increase in Nusselt number by 52%, the convective heat transfer coefficient by 49% and the overall heat transfer coefficient by 68%. Farajollahi et al. measured thermal transfer properties of water/Al₂O₃ and water/TiO₂ nanofluids in a shell and tube heat exchanger under turbulent flow conditions [4]. They studied the effect of Peclet number and nanofluid concentrated on heat transfer characteristics. Based on the results, the heat transfer behavior is generally improved. Also, heat transfer characteristics of

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TiO₂ nanofluid with an optimum concentration (0.3 vol. %) and a certain pecelet number (20000-60000) was higher than Al₂O₃ nanofluid. Anoop et al. reported the effect of SiO₂/water nanofluid on the overall heat transfer coefficient and pressure drop in plate and shell and tube heat exchangers [5]. The results revealed that heat transfer coefficient was a function of flow rate and nanofluid concentration in both heat exchangers. Albadr et al. investigated the effect of Al₂O₃ concentration on the convective heat transfer and friction factor in a shell and tube heat exchanger under counter-current and turbulent flow conditions [6]. The results indicated that the heat transfer coefficient was enhanced by increasing the nanofluid concentration and mass flow rate. Lotfi et al. studied the effect of MWCNT/water nanofluid heat transfer in a shell and tube heat exchanger [7]. The results showed an increase in the heat transfer coefficient compared with the base fluid (water). Zeinali Heris et al. compared the convective heat transfer of CuO/water and Al₂O₃/water nanofluids through a circular tube under laminar flow conditions with constant wall temperature [8]. The results indicated that the increase in nanofluid concentration and pecelet number increased the heat transfer coefficient. Also, the enhanced heat transfer of Al₂O₃/water nanofluid was higher than that of CuO/water. Sajadi et al. reported the heat transfer and pressure drop characteristics of ZnO/water nanofluid in a circular tube heat exchanger under constant wall temperature conditions [9]. The results indicated that nanoparticle concentration can increase heat transfer coefficient, overall thermal performance and pressure drop. Tasi et al. investigated the heat transfer of Au/water nanofluids [10]. Experimental results demonstrated that under identical conditions, nanofluid results in a significant reduction in thermal resistance of tubing compared to the base fluid. Ghozatloo et al. studied the enhancement of water/graphene nanofluid heat transfer in a shell and tube heat exchanger under laminar flow conditions [11]. The results indicated that the convective heat transfer coefficient was enhanced by increasing nanofluid concentration and temperature. Godson et al. studied the heat transfer of Ag/water nanofluid in a shell and tube exchanger [12]. They studied the influence of inlet temperature, Reynolds number, nanofluid concentration and mass flow rate on LMTD, effectiveness, convection heat transfer and pressure drop. The results showed that enhancement in the heat transfer coefficient, pressure drop, LMTD and effectiveness is performed

by increasing nanofluid concentration and Reynold's number.

Based on the results reported in the literature, TiO₂ is a promising material for heat transfer improvement which is previously studied as a photo-catalyst [13, 14] and membrane material both as support [15] and the top-layer [16]. So, this study is aimed at the improvement of TiO₂ in a nanocomposite structure with ZnO which is also a promising material for heat transfer coefficient enhancement. ZnO/TiO₂ nanocomposite is synthesized through various techniques and evaluated in several applications such as solar cells [17], photo-catalysts [18], sensors [19] and many others. In this study, sol-gel and mechanical techniques are implemented for the synthesis and preparation of ZnO/TiO₂ nanocomposite. Corresponding nanofluids were prepared and used in a shell-tube heat exchanger for investigating the effect of concentration on the heat transfer and overall heat transfer coefficient at various temperatures. It is expected that ZnO/TiO₂ nanocomposite can be a good choice for heat transfer enhancement in tube heat exchangers.

EXPERIMENTAL

Materials and methods

TiO₂ nanoparticles were purchased from Tecnan Co. The average particle size and specific surface area was 10-15 nm and 100-150 m²/g, respectively. Zinc acetate, isopropyl alcohol, NaOH, ammonium hydroxide, ethanol and methanol were purchased from Merck Co. All solutions were prepared using deionized water.

Synthesis of ZnO nanoparticles

ZnO nanoparticles were synthesized through precipitation technique. 0.1 g of zinc acetate and isopropyl alcohol were mixed at 65 °C for 15 min by magnetic stirrer. Then the mixture was added to 12 ml of isopropyl alcohol placed in a water bath at 0 °C. 15 ml of NaOH (0.05M) in the water bath (0°C) was added drop-wise into the final solution and stirred at a high stirring rate. The produced solution was maintained under 65 °C for 5min. Finally, it was washed using ethanol, methanol and water 3 times and then cooled at the ambient temperature [20].

Synthesis of ZnO/TiO₂ nanocomposite

Preparation of ZnO sol

1ml of ammonium hydroxide was mixed with 100ml of zinc chloride (0.2 M) until a white precipitate of zinc hydroxide was formed. Adding

more ammonium hydroxide (1 ml) leads to the dissolution of the precipitate [21].

Preparation of TiO₂ sol

1 ml of titanium isopropoxide was dissolved into isopropyl alcohol (32ml) and mixed with magnetic stirring for 2 h. 0.05 ml of citric acid was added to the final solution to prepare a uniform sol [13, 14]. ZnO and TiO₂ solswere mixed to obtain the nanocomposite in a 50-50 molar ratio and then sonicated for 5 min. For drying the ultimate product, it was kept at room temperature for 72 h. At last, the prepared nanocomposite was calcined at various temperatures (500 °C, 600 °C & 800°C) for 1 h. TiO₂/ZnO nanocomposites with 25-75 mol%, 50-50 mol%, 75-25 mol% proportions were obtained by mixing appropriate proportions of the as-prepared sols.

Preparation of ZnO/TiO₂ nanocomposite by mechanical technique

In order to prepare the nanocomposite by mechanical method, ZnO and TiO₂ nanoparticles with 25-75 mol%· 50-50 mol% and 75-25mol% were mixed with ethanol for 15 min by a high speed magnetic stirrer and then sonicated for 30 min. The mixture was filtered and washed with water and ethanol 3 times and dried at room temperature for 24 h. Finally, the obtained nanopowder was sintered at 550°C (with a 4.7 C/min. rate) for 2 h.

Preparation of ZnO and TiO₂ nanofluids

To prepare ZnO and TiO₂ and ZnO/TiO₂ nanofluids, nanoparticles with 0.001, 0.01 and 0.1 wt. % were added to deionized water (2:1) and stirred for 15-20 min. At last, the mixture was sonicated (100 W) for 45 min to disperse the nanoparticles in the base fluid.

Heat exchange experiments

To study the effect of nanofluid on heat exchange characteristics, an experimental setup was designed. The heat exchanger is a shell and tube type with counter-current flow, including both heating and cooling loops. The nanofluid flowing through the shell consists of a coolant roll and deionized water through the tubing was the heating medium. The heating loop consists of a pump, hot fluid tank, heater, temperature controller and measurement of tube temperature inlet and outlet systems. The cooling loop consists of a pump, nanofluid reservoir tank, temperature controller and the measurement of shell temperature inlet and outlet systems. The shell and tube were made of pyrex and copper, respectively. The details of the system can be found elsewhere [22]. To measure the heat transfer coefficient (h) and the overall heat transfer coefficient (U), experiments were carried out in this heat exchanger with counter-current flow at two temperatures (50 °C and 60 °C) for water/water, water/nanoparticle and water/nanocomposite systems.

RESULTS AND DISCUSSION

Characterization

XRD Analysis

The XRD pattern of ZnO/TiO₂ nano-composite is given in Fig. 1. The sharp diffraction peaks show the nano-crystalline structure. Sharp characteristic peaks are observed at $2\theta = 25.5, 27.5, 32.5, 35.5, 41.5, 44.5, 48.5$ and 57 for the ZnO-TiO₂ nanocomposite [23-25]. Compared with XRD patterns for ZnO and rutile TiO₂ phase, whole peaks are shifted before $2\theta = 60$ while higher 2θ peaks are also observed in ZnO and TiO₂ separately. The characteristic peaks of each phase are also seen in the nanocomposite while other peaks disappeared which show a change of morphology in the nanocomposite.

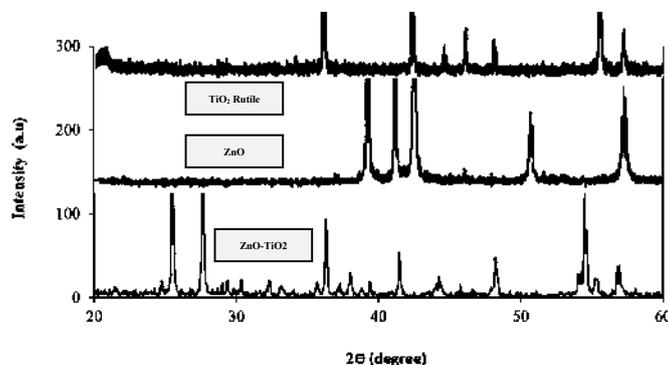


Fig.1. XRD patterns for ZnO/TiO₂ (50/50 mol %) nano-composite, ZnO and TiO₂ (Rutile).

Particle size analysis (PSA)

The ZnO/TiO₂ nano-composite is characterized by particle size analyzer (Horiba LB-360) (0.001-6µm). The PSD of TiO₂/ZnO nano-composite is given in Fig. 2 which shows an average particle size of 40nm with a uniform size distribution.

FTIR analysis

The FTIR spectrum of the synthesized nanocomposite was recorded in the range of 400–4000 cm⁻¹ shown in Fig. 3. The peaks at 3600 cm⁻¹ and 1607 cm⁻¹ were attributed to OH stretching and bending of water, respectively. The band at around 1406 cm⁻¹ was attributed to the vibration

mode of Ti–O bond. The absorption peaks at ~ 600 cm⁻¹ are also related to Zn-O bonding. The peak at 974 cm⁻¹ is responsible for Ti-O bonds [26].

SEM micrograph

Fig. 4 represents the SEM micrograph of the ZnO/TiO₂ nano-composite. Based on the Figure, semi-spherical nanoparticles are observed in the ~ 100 nm size range and smaller.

Heat Exchange Experiments

Effect of nanofluid concentration

Figs. 5 and 6 show the effect of TiO₂ and ZnO nanofluid concentration on the heat transfer

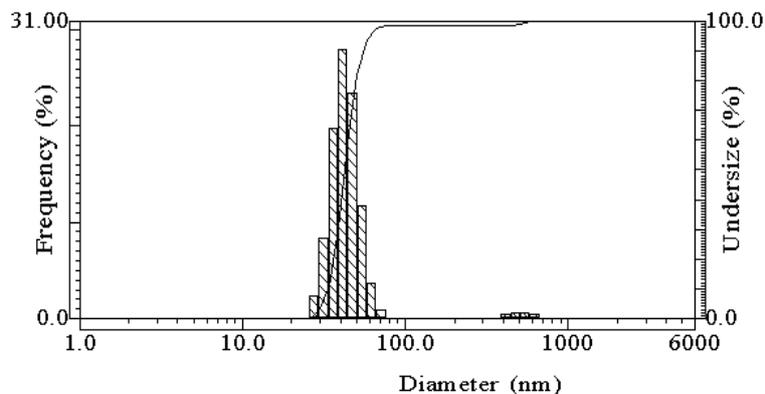


Fig. 2. PSD of ZnO/TiO₂ (50/50 mol %) nano-composite.

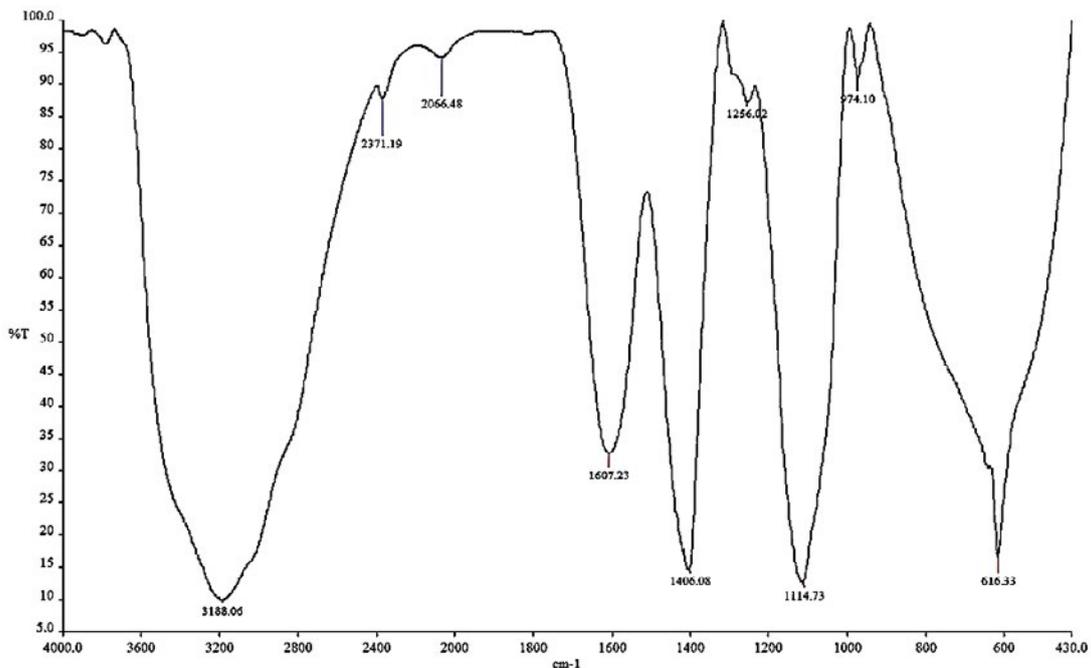


Fig. 3. FTIR spectrum of ZnO/TiO₂ nano-composite.

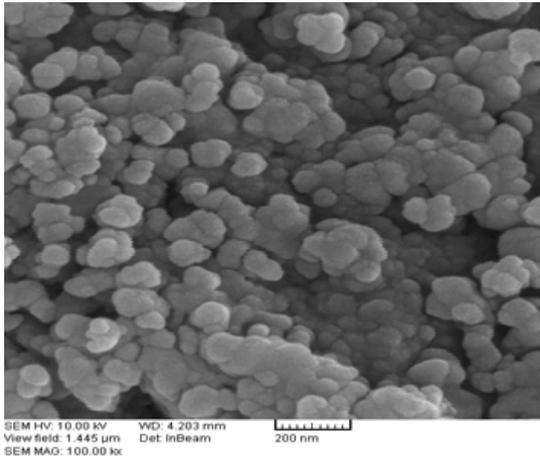


Fig. 4. SEM micrograph of ZnO/TiO₂ nano-composite.

coefficient and the overall heat transfer coefficient in the heat exchanger at 50°C. The results indicate that the heat transfer coefficient is increased by increasing the nanofluid concentration due to the enhanced thermal conductivity and reduced specific heat capacity and thermal boundary layer thickness. The heat transfer coefficient and the overall heat transfer coefficient are 0.5865 W/m²°C and 0.5895 W/m²°C for the water/water system, respectively. A significant increase

is observed using nanofluids in the shell side. The heat transfer coefficient of TiO₂ nanofluid is usually higher than that of ZnO, except for 0.1 wt. % which may be due to the high stability of ZnO at higher concentrations. It must be mentioned that nanofluid stability is a function of several factors such as size, size distribution, solvent type, temperature, zeta potential and many others which must be calculated per case. Usually, heat transfer coefficients are enhanced by adding stable nanoparticles to the base fluid and this continues up to the concentration that instabilities are observed which results in the reduction of heat transfer coefficient thereupon. The same trend is also observed for the overall heat transfer coefficient.

Table 1 shows the heat transfer coefficient and overall heat transfer coefficient enhancement compared with the base (water/water) fluid. The maximum heat transfer coefficients occurred at 0.01 wt. % of ZnO and TiO₂ in their corresponding nanofluids. Higher than 0.01 wt.% a slight enhancement occurs in the heat transfer coefficient in both nanofluids which can be ignored due to economic reasons. Experiments performed higher than 0.1 wt. % show that heat transfer coefficient reduction is observed due to agglomeration.

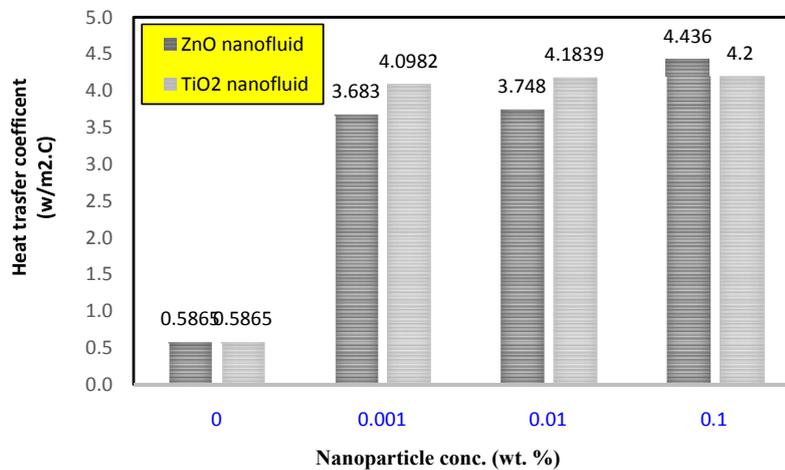


Fig. 5. Effect of TiO₂ and ZnO nanofluid concentration on the heat transfer coefficient.

Table 1. Heat transfer coefficients enhancement.

nanofluid conce (wt. %)	Heat transfer coefficient (W/m ² °C)		Overall heat transfer coefficient (W/m ² °C)	
	TiO ₂ nanofluid	ZnO nanofluid	TiO ₂ nanofluid	ZnO nanofluid
1	0.001	5.99	3.9	2.98
2	0.01	6.13	4.05	3.08
3	0.1	6.16	4.26	4.36

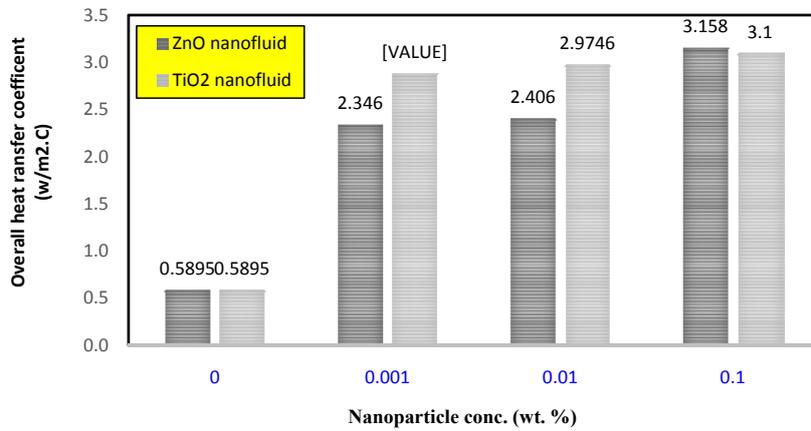


Fig. 6. Effect of TiO₂ and ZnO nanofluid concentrate on the overall heat transfer coefficient.

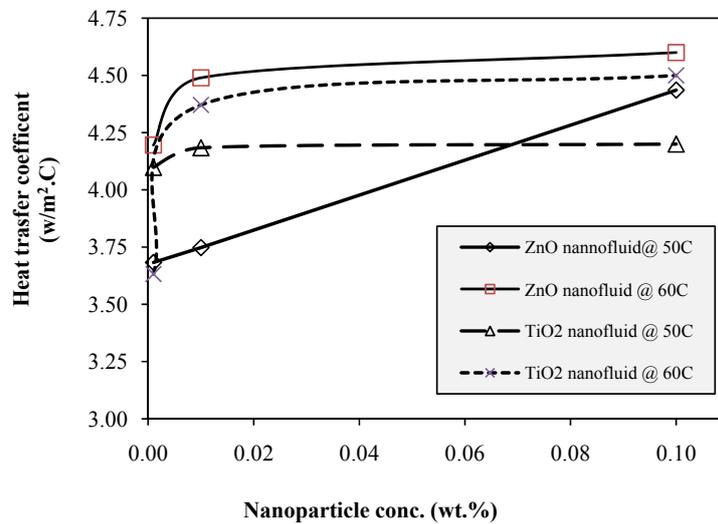


Fig. 7. Effect of temperature on the heat transfer coefficient.

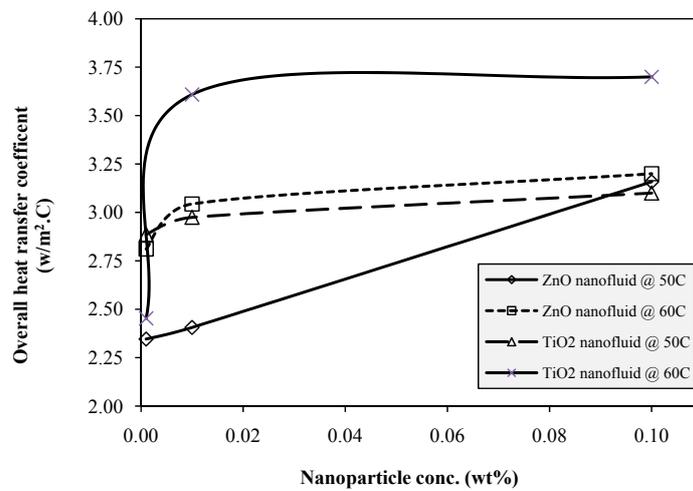


Fig. 8. Effect of temperature on the overall heat transfer coefficient.

Table 2. Heat transfer coefficient enhancement compared to the base fluids.

ZnO/TiO ₂ ratio	Heat transfer coefficient (W/m ² °C)			Overall heat transfer coefficient (W/m ² °C)		
	25-75%	50-50%	75-25%	25-75%	50-50%	75-25%
Nanofluid conc. (wt. %)						
0.001	4.74	3.79	4.33	7.38	4.27	6.48
0.01	6.05	4.5	4.98	8.68	4.75	6.95
0.1	9.09	4.68	5.56	9.14	4.83	6.99

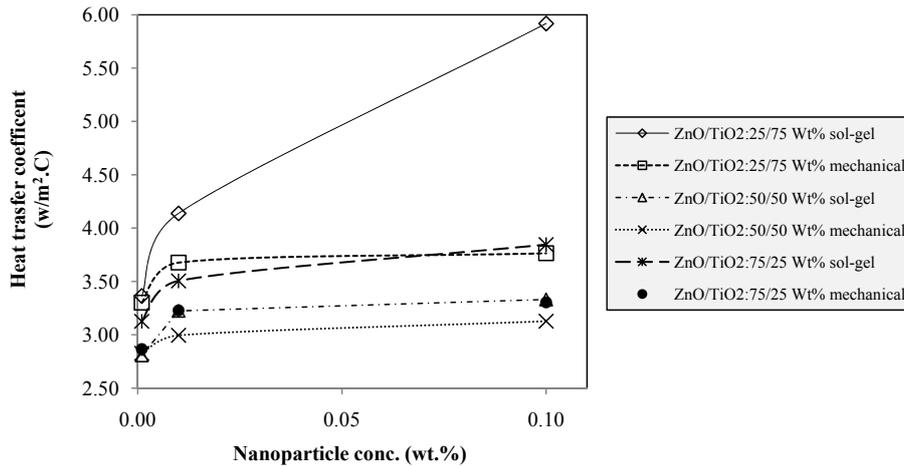


Fig. 9. Effect of ZnO/TiO₂ nano-composite on heat transfer coefficient.

Effect of temperature

The effect of temperature on heat transfer coefficient and overall heat transfer coefficient are shown in Figs. 7 and 8. As the temperature is increased, the heat transfer coefficient is raised at all nanofluid concentrations. Heat transfer coefficient variations of TiO₂ and ZnO nanofluid in comparison with the concentrations at 60 °C is higher than that of 50 °C. The variations of heat transfer coefficient of ZnO nanofluid are higher than that of TiO₂ nanofluid.

The heat transfer coefficient enhancement compared to the original system (water/water) is given in Table 2. The maximum heat transfer coefficient of nanofluid is occurred at 25/75 mol% of ZnO/TiO₂ nano-composite.

Effect of nanocomposite concentration

The heat transfer coefficient and the overall heat transfer coefficient of the nanofluids obtained from the nanocomposite prepared by sol-gel and mechanical techniques with 75-25%, 50-50% and 25-75% of ratios of TiO₂/ZnO are illustrated in Figs. 9 and 10, respectively. The results reveal that by increasing the nanocomposite concentration,

heat transfer and overall heat transfer coefficients are increased, which is apparently higher than that of ZnO and TiO₂ nanoparticles separately. Based on the results, the heat transfer coefficients of nanofluids obtained from the nanocomposite by sol-gel is higher than that of the prepared nanocomposite by mechanical technique which is possibly due to higher crystallinity of the product from precipitation compared with the mechanical method.

CONCLUSIONS

In this work, ZnO nanoparticle and ZnO/TiO₂ nanocomposite were synthesized via precipitation and sol-gel techniques. The nanofluid is prepared and characterized by PSA, XRD, FTIR and FE-SEM analyses. Effect of nanofluid concentration and temperature on heat transfer coefficient and overall heat transfer of the nanofluid in shell-tube heat exchanger are investigated. The results show that:

- The heat transfer and overall heat transfer coefficients of nanofluids are higher than that of the base fluid enhanced by nanofluid concentration.
- The nanofluid heat transfer coefficient variations



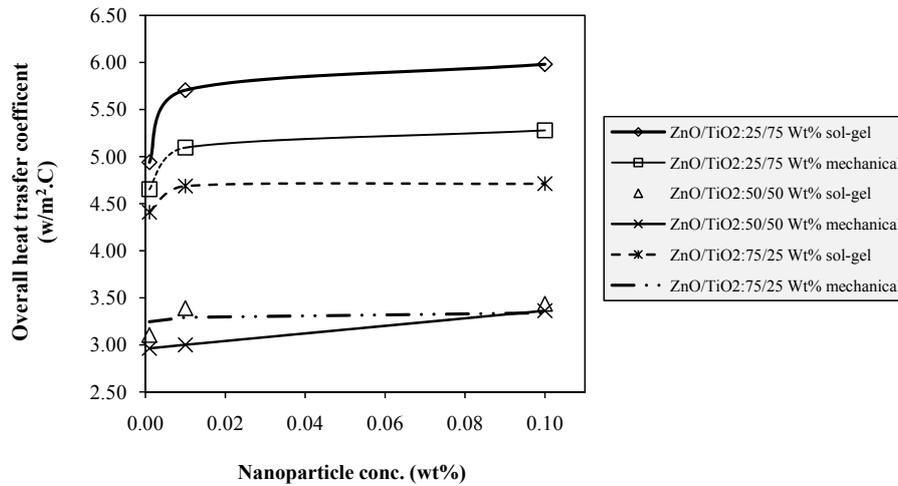


Fig. 10. Effect of ZnO/TiO₂ nano-composite on the overall heat transfer coefficient.

at 60 °C are higher than that of 50 °C.

- The heat transfer coefficients of TiO₂ nanofluid are higher than that of ZnO but the heat transfer coefficients variations of ZnO nanofluid are higher than that of TiO₂.

- The heat transfer coefficients of nanocomposites are higher than that ZnO and TiO₂ nanofluids and the base fluid. The highest heat transfer coefficients occur at the ZnO/TiO₂ = 25/75% molar ratio.

CONFLICT OF INTEREST

The authors declare that there is no conflict of interests regarding the publication of this manuscript.

REFERENCES

- [1] G. Humnic and A. Humnic, *Renew. Sustain. Energy Reviews*, 16, 5625 (2012).
- [2] S.S. Sonawane, R. S. Khedkar and K.L.Wasewar, *Int. Commun. Heat Mass Trans.*, 49, 60 (2013).
- [3] D. Madhesh, R. Parameshwaran and S. Kalaiselvam, *Experimental Thermal and Fluid Sci.*, 52, 104 (2014).
- [4] B. Farajollahi, S.Gh. Etemad and M. Hojjat, *Int. J. of Heat Mass Transfer*. 53, 12 (2010).
- [5] K. Anoop, J. Cox and R. Sadr, *Int. Commun. Heat and Mass Transfer.*, 49, 5 (2013).
- [6] J. Albadr, S. Tayal and M. Alasadi, *Case Studies in Thermal Eng.*, 1, 38 (2013).
- [7] R. Lotfi, A. M. Rashidi and A. Amrollahi, *Int. Commun. Heat and Mass Transfer.*, 39, 108 (2012).
- [8] S. Zeinali-Heris, M. Nasr Esfahany and G. Etemad, *J. Enhanced Heat Transfer.*, 13, 279 (2006).
- [9] A.R. Sajadi, S.S. Sadati, M. Nourimotlagh, O. Pakbaz, D. Ashtiani and F. Kowsari, *Thermal Sci.*, 18, 1315 (2014).
- [10] P. Gurav, S.S. Naik, K. Ansari, S. Srinath, K. A. Kishore, Y. P. Setty and S. Sonawane, *Colloids Surfaces A: Physicochem. Eng. Aspects.*, 441, 589 (2014).
- [11] A. Ghozatloo, A. Rashidi and M. Shariaty-Niassar, *Exp. Thermal Fluid Sci.*, 53, 136 (2014).
- [12] L. Godson, K. Deepak, C. Enoch, B. Jefferson and B. Raja, *Archives Civil Mech. Eng.*, 14, 489 (2014).
- [13] S. Sabbaghi, H. Sadeghi, M. Mohammadi, M.M. Zerafat and R. Pooladi, *Desal. Water Treatm.*, 57, 799 (2016).
- [14] M. Mirzayi, S. Sabbaghi and M.M. Zerafat, *Canadian J. Chem. Eng.*, 96, Article in Press (2018).
- [15] M.H. Yousefi, M.M. Zerafat, M. Shok-Doodeji and S. Sabbaghi, *J. Water Environment. Nanotech.*, 2, 235 (2017).
- [16] M. Shokri-Doodeji, M.M. Zerafat and O. Nejadian, *J. Nanoanalysis*, 4, 247 (2017).
- [17] A.E. Shalan, A. Mourtada, M. Rasly, M.M. Moharam, M. Lira-Cantu and M.M. Rashad, *RSC Advances*, 5, 103095 (2015).
- [18] C. Karunakaran, G. Abiramasundari, P. Gomathisankar, G. Manikandan and V. Anandi, *Materials Research Bulletin*, 46, 1586 (2011).
- [19] B.C. Yadav, Richa Srivastava and C.D. Dwivedi, *Philosophical Magazine*, 88, 1113 (2008).
- [20] S.S. Kanmani and K. Ramachandran, *Renewable Energy*, 43, 149 (2012).
- [21] M.R. Vaezi, *J. Materials Process. Tech.*, 205, 332 (2008).
- [22] T. Namdar, M.M. Zerafat, S. Sabbaghi, R. Saboori and M. Yousefifar, *Transp. Phenom. Nano. Micro. Scales*, 6, 104 (2018).
- [23] M. Zhang, T. An, X. Liu, X. Hu, G. Sheng and J. Fu, *Materials Lett.*, 64, 1883 (2010).
- [24] L. Wang, X. Fu, Y. Han, E. Chang, Haitao Wu, H. Wang, K. Li and X. Qi, *Hindawi Publishing Corporation J. Nanomaterial*. 2013, 6 (2013).
- [25] M.A. Habib, M.T. Shahadat, N.M. Bahadur, I.M.I. Ismail and A.J. Mahmood, *Int. Nano Letters.*, 3, 5 (2013).
- [26] N.M. Bahadur, T. Furusawa, M. Sato, F. Kurayama, N. Suzuki, *Materials Research Bulletin*. 45, 1383 (2010).